Chemistry Revision Test Time: 60 mins Max Marks: 30

Q1.

- (a) Name the structure for the following compound.2-(2-chlorophenyl)-1-iodooctane, p-Bromochlorobenzene
- (b) Arrange the following compounds in ascending order of dipole moments. CH_2Cl_2 , $CHCl_3$, CCl_4
- (c) How you will convert but-1-ene to 1-lodobutane?
- (d) Explain Ambient nucleophiles. Give one example4 marks
- Q2. A) Write the structures of main products in reaction below:
 - (i).2 Chloro -2 Methyl butane with Sodium ethoxide in ethanol.
 - (ii). 2,2,3-Trimethyl-3-bromopentane with sodium ethoxide in ethanol.
 - B) Complete the following reactions:
 - (i).C2H₅Cl +C₂HNa ----> ?
 - (ii) CH₂=CH₂-CH₃+HBr ---Peroxide-- >?

4 marks

Q3.

Explain why:

- (a). Alkyl halides, though polar are immiscible with water.
- (b).p-Dibromobenzene has higher melting point than those of o and m isomers?
- (c).Chloro Diphenylmethane is hydrolysed faster than Benzyl chloride?

3 marks

- B) What happens when?
- (a). Toulene is treated with Chlorine in UV light
- (b).Benzene with Bromine water in presence of FeBr₃?
 - . 3 marks
- C) How you will convert:

- (a).Propene into Propane-1-ol
- (b). Ethanol to Propanenitrile
- (c).Aniline to Chlorobenzene.
- 3 marks
- Q4 Give reasons:
- (i).Alkyl halides when treated with aqueous KOH forms alcohols but with alcoholic
- KOH forms alkene.
- (ii).Sulphuric acids not used in the reaction of alcohols with KI.
- (iii).Freon-12 is banned in many countries.

3 marks

Q5

Identify A, B, C, D, E, R and R¹ in the following:

5 marks

Q6.Give reasons:

(i).Tert-Butyl bromide undergoes Wurtz reaction to give 2,2,3,3-tetramethlybutane.

(ii).Presence of nitro group and ortho and para position increases the reactivity of

haloarenes towards nucleophilic substitution.

(iii).Chlorine in chlorobenzene is difficult to replace than in chloroethane by OH group.

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(iv). Some alkyl halides undergo substitution whereas some undergoes elimination on treatment with bases.

(v). What are the IUPAC names of the insecticide DDT and benzenehexachloride?5 marks